International Journal of Research

Available at https://edupediapublications.org/journals

p-ISSN: 2348-6848 e-ISSN: 2348-795X Volume 03 Issue 09 May 2016

Green Synthesis and Characterization of Gold Nanoparticles from *Erythrina Variegate* **Plant Extract**

Abdulsalam Mirdin ali; Hassanain Hataf Jaber

Masters of Science in Nano Biotechnology, Department of Nano-biotechnology, Acharya Nagarjuna University

Jaafar Badr Obald

Masters of Environmental Science, Department of Environmental Science, Acharya Nagarjuna University

Abstract:

To develop a reliable, eco-friendly and easy process for the synthesis of gold nanoparticles using leave extracts of medicinal plant "Erythrina Variegate". The use of plant material not only makes the process eco-friendly but also the abundance makes it more economical. In the present research program, cost effective and environment friendly gold nanoparticles were synthesized using the Erythrina Variegateleave extract as the reducing agent. The biosynthesized gold nanoparticles were analysed by UV- Vis,FTIR, XRD, EDX, SEM and TEM. The broadening of peaks in the XRD patterns was attributed to particle size effects. On whole, the aunps obtained by the biogenic syntheses have potential biological and medical applications. UV-visible spectrum of the aqueous medium containing gold nanoparticles showed a peak around 537nm. FT-IR analysis confirmed reduction of Au3+ ions to Au0 ions in synthesized gold nanoparticles. The leave extract of "Erythrina Variegate" quickly reduces Au3+ to Au0 and enhances the synthesis of gold nanoparticles.

INTRODUCTION

Nanomaterials have received much attention because of their structure and properties differ significantly from those of atoms, molecules, and bulk material.[1] The synthesis of metal nanoparticles has been widely discussed in the literature due to their small sizes, large surface area and unique physical and chemical properties, which have many potential applications.[2-4] Biosynthesis of nanoparticles is budding into a significant approach in nanotechnology.[5,6]

Green chemistry approach method highlights the usage of natural organisms as nontoxic, ecofriendly, reliable and simple.[7,8] Synthesis of nanoparticles using microorganisms, enzyme and plant extract, have been proposed by several researchers[9,10]. Gold nanoparticles have wide range of application which includes biomedical imaging, cancer therapy, diagnosis, biological and chemical sensing. Green synthesis of gold nanoparticles using plants extract such as Neem [11-13], Alfalfa, Emblicaofficinalis, Cinnamomumcamphora, Hibiscus rosasinensis,

Camellia Sinensis, Coriander leaves, Sugar beet pulp Menthapiperita and AegleMarmelos [14-22] have been reported.

"Erythrina Variegate" belonging to the family Fabaceae is commonly known as Tiger claw. Erythrina Variegate is a most important species used to treat several ailments in traditional "Erythrina medicine.[23-26] system of Variegate" was reported for its antidiarrhoeal, anticancer[27-29], thyroid gland stimulating properties. Since, the plants are rich in phytoconstituents, it was considered as a principle source in pharma and nutraceutical industries.. It has been known to possess antibacterial, antidiabetic, analgesic, antiinflammatory, nephroprotective, and thyroid hormoneregulating activity.[30]

It is a thorny deciduous tree growing to 27 m (89 ft) tall. The leaves are pinnate with a 20 cm petiole and three leaflets, each leaflet up to 20 cm long and broad. It has dense clusters of scarlet or crimson flowers and black seeds.[31-32]It is valued as an ornamental tree. Several cultivars have been selected, including 'Alba'

International Journal of Research

Available at https://edupediapublications.org/journals

p-ISSN: 2348-6848 e-ISSN: 2348-795X Volume 03 Issue 09 May 2016

with white flowers.[33]It was designated the official flower of Okinawa Prefecture in 1967. The deigo flower features in the popular song "ShimaUta" by The Boom, one of the most wellknown songs associated with Okinawa. In addition, the use of the wood of the deigo tree is of the unique characteristics one Ryukyuanlacquerware.[34-35] In our present study, we have demonstrated a suitable green route for the synthesis and characterisation of gold nanoparticles using "Erythrina Variegate" leave extract as reducing agent.[36]

MATERIALS AND METHODS Materials

Chloro Auric acid (HAuCl4) was purchased from National Scientfic Company,

Madurai, India and used as received. Double distilled (DD) water was used for experimental work. Erythrinavariegata leaves were collected from S.F.R College for Women, Sivakasi, Virudhunagar district.

Preparation of Erythrinavariegata leaves extract:

Fresh leaves of Erythrinavariegate washed several times with water to remove the dustparticles and then dried to dark condition to removethe residual moisture and grinded to form powder. 10g Erythrina variegate leaves powder is taken in abeaker and added 100ml of distilled water. Then it isheated and stirred in magnetic stirrer. It is allowed for 30 mins. The Erythrina variegate solution is filtered in ice cold condition and the plant extract is prepared. The extract is stored at room temperature and is used for the biosynthesis of gold Nps.

Synthesis of Gold NPs:

In the typically synthesisprocess of gold NPs, add 2ml of Erythrina variegatedried leaf powder extract into the 8 ml of goldchloride solution under stirring at 600c temperature. The bio reduced component was confirmed by using UV-Visible spectrophotometer. A reduction of AuNps was clearly observed when Erythrina variegateleaves extract was added with gold chloride solution within 10 mins. The dark brown solution changed toruby red colour which indicates the formation of gold Nps.

Characterization techniques

UV-visible absorption spectrum was measured using Shimadzu UV-2400 spectrophotometer, Japan. Crystalline metallic gold NPs were examined using an X-ray diffractometer equipped with using Cu-K α radiation (λ = 1.54060 Å) at 40 kV over the range of $2\theta = 10^{\circ}$ -90° with a step size of 0.020 and step time of 160.0s. Phase purity and grain size are determined by XRD analysis. Fourier transform infrared (FTIR) spectra for Erythrina Variegate leave extract end capped gold NPs was obtained the range 4000 to 400 cm-1 KBrpelletization method by using Shimadzu 8400S, Japan model instrument. Scanning electron microscopy (SEM) with Energy Dispersive X-Ray Spectrometer (EDS) analysis of synthesized gold NPs was done using a SEM CARL ZEISS EVO 18 and EDS Quantax 200 with X-Flash – Bruker machine. The morphology of the Au NP was analyzed with the TEM (Model JEOL - 2100EX).

Available at https://edupediapublications.org/journals

p-ISSN: 2348-6848 e-ISSN: 2348-795X Volume 03 Issue 09 May 2016

RESULTS AND DISCUSSION

UV-visible spectral analysis:

The mixture of leave extract and HAuCl4 solution was subjected to UV-visible spectroscopy analysis and it shows a peak at 537 nm corresponds to the wavelength of the Surface Plasmon Resonance of Au NPs (**Figure 1**). Various reports have established that the resonance peak of Gold NPs appears around this region.

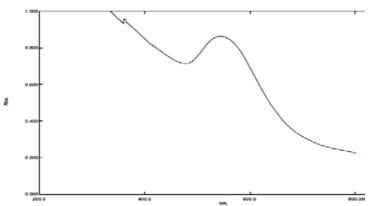


Figure 1: UV-visible spectrum of Erythrina Variegate plant extract decorated Gold Nano Particle.

FTIR spectroscopy:

FTIR analysis was performed to identify the possible functionalities of biomolecules responsible for the reduction of the Au+ ions and capping of the reduced AuNPs synthesized using *Erythrina Variegate* leave extract (**Figure2**) showed strong bands at 3379, 2924, 1643, 1550, 1041, 833,763,663,524 and 439 cm-1. Bands appeared at 3350 and 2924 cm-1 are corresponding to O-H stretching and aliphatic - C-H stretching respectively. The bands at 1651 and 1041cm-1 due to the C=C stretching and C-O-C stretching respectively. Hence, the main components such as steroids, saponins, tannins, Phenolic compounds, flavonoids, and glycosides present in the leave extract of *Erythrina Variegate plant* are prime responsible for the observed reduction and capping during the synthesis of AuNPs. The two strong bands recorded at 763 and 439 cm-1 in the spectrum of the synthesized material were assigned to the C-H bending and metal-oxide (Au-O), respectively. The C-H bending peak is due to the reduction of HAuCl4 to Au NPs by the leave extract.

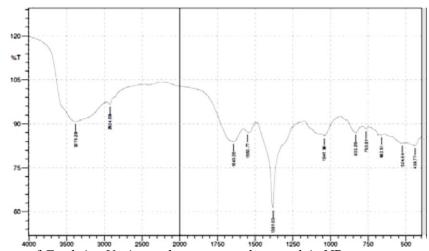


Figure2:FTIR spectrum of Erythrina Variegate leave extract decorated AuNP

p-ISSN: 2348-6848 e-ISSN: 2348-795X Volume 03 Issue 09 May 2016

X-ray diffraction analysis

X-ray diffraction (XRD) pattern was recorded for the synthesized Au NPs is given in (**Figure 3**). Three distinct diffraction peaks at 38.73°, 45.03°,65.41° and 78.26 were indexed with the planes d111, d200, d220 and d311and these peaks are matched with the face-centered cubic (fcc) structure of gold as per the JCPDS card no.011174. The well resolved and intensed XRD pattern clearly showed that the Au NPs formed by the reduction of Au+ ions using *Erythrina Variegate*leave extract are crystalline in nature. The results were compared with the literature and found to be matched. The low intense peak at 78.26° belongs to d311 crystal plane.

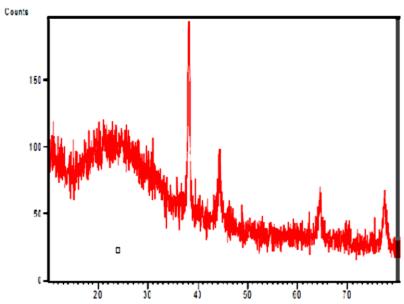


Figure 3: XRD pattern of Erythrina Variegateleave extract decorated AuNP

EDX analysis

Energy-dispersive X-ray analysis (EDX) was performed to determine the elemental composition of the AuNPs synthesized and stabilized by the *Erythrina Variegate*leaveextract. Strong peaks of Au in the area-profile analysis of the synthesized nanoparticles (Fig.4) confirm the formation of AuNPs.

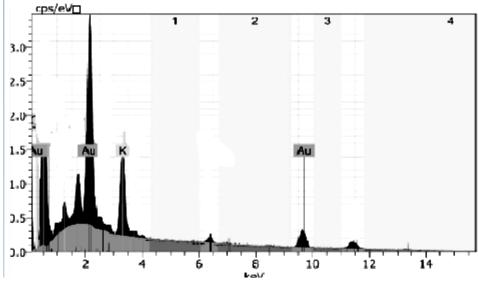


Figure4.EDAX profile of AuNPs synthesized from Erythrina Variegate plant extract.

Available at https://edupediapublications.org/journals

p-ISSN: 2348-6848 e-ISSN: 2348-795X Volume 03 Issue 09 May 2016

SEM analysis

The SEM image (**Figure 5**) of plant extract decorated Au NP exhibited the distorted cubical morphology with agglomerated structure. The size was determined as 20-50 nm. The important point noted here is polydispersity of size and shape is minimised.

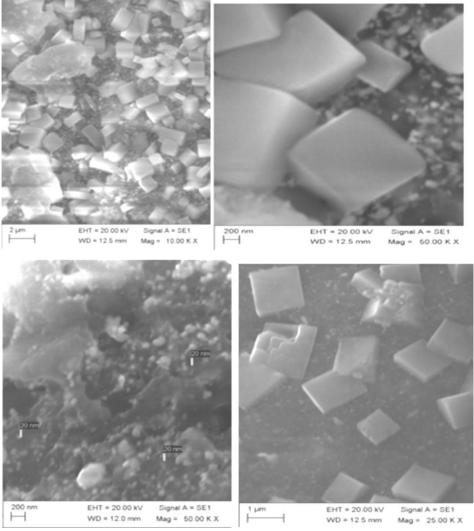


Figure 5: SEM image of Erythrina Variegateleaveextract decorated Au NP

TEM Measurements

TEM measurements were performed on a JEOL model 2100EX instrument operated at an accelerating voltage at 200 kV TEM. A TEM study reveals the size and shape of nanoparticles. The shape of gold nanoparticles prepared in this study is cubic with size in the range of 20–50 nm (Figure 6). Some nanoparticles are seen as aggregated. In the surface of the nanoparticles some light colour of materials found that may be the stabilizing agent responsible for the nanoparticles synthesis. The separation between nanoparticles was observed from TEM image due to the presence of capping agent which is characteristics to well dispersed nanoparticles formation in the optimized conditions.

International Journal of Research

Available at https://edupediapublications.org/journals

p-ISSN: 2348-6848 e-ISSN: 2348-795X Volume 03 Issue 09 May 2016

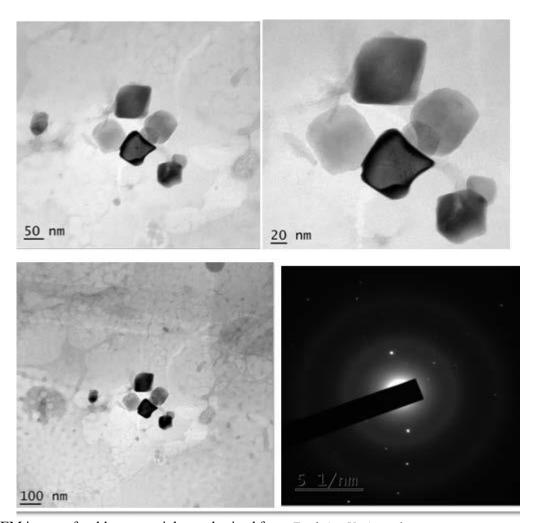


Figure6. TEM image of gold nanoparticle synthesized from Erythrina Variegateleaveextract

CONCLUSIONS

The reduction of gold ions to gold nanoparticles was achieved using the aqueous extract of Erythrina Variegate plant. The nanocube produces many actives sites which may be exploited in organic synthesis or in reactions for enhanced catalytic activity. Appearance of a peak at 537 nm confirmed the presence of extract decorated Au NP. FT-IR study showed absorption bands corresponding to the main functional groups present in natural plant extracts. The average particle size of AuNPs was found to be between 20 nm - 50 nm. The AuNPs were crystalline as revealed from XRD studies. The average size of the AuNPs was 20 nm as deter-mined with Debye-Scherrer's equation utilizing XRD data TEM declared the Cubic shaped morphology of Au NP. SEM also confirmed the distorted morphology of Au NP. The present protocol is an eco-friendly and easy way of gold nanoparticles synthesis under laboratory conditions.

REFERENCES

- [1] Rosei F, "Nanostructured surfaces: challenges and frontiers in nanotechnology," *Journal of Physics Condensed J PhysCondens Matter*, (2004),16, pp.S1373–S1436.
- [2] Papp S, Patakfalvi R, Dekany I " Metal nanoparticle formation on layer silicatelamellae." *Colloid PolymSci*, (2008), 286,pp.3–14.
- [3] Seoudi R., Said R A, "Studies on the effect of the capping materials on the spherical gold nanoparticles catalytic activity." WJNSE,(**2011**), 1.11,pp. 1: 51-61.
- [4] Foglia S, Suber L, Righini M. "Size tailoring of CdS nanoparticles by different colloidal chemical techniques". *Colloids Surf*, **(2001)**, 177(1),pp.3-12.



Available at https://edupediapublications.org/journals

p-ISSN: 2348-6848 e-ISSN: 2348-795X Volume 03 Issue 09 May 2016

- [5] Raveendran P, Fu J Wallen SL. "Synthesis of silver nanoparticles using Cossiaaurticulata leaf extract" *A simple and Green Chem*, J Am Chem(**2006**), 46,34.
- [6] Magudapathy P, Gangopadhyay P, Panigrahi BK, Nair KGM, Dhara S, "Electrical transport studies of Ag nanoclusters embedded in glass matrix". *Physica B*, (2001),299,142-146.
- [7] Sathishkumar M, Sneha K, Yun YS. "Immobilization of silver nanoparticles synthesized using curcuma longa tuber powder and extract on cotton cloth for bactericidal activity". *BioresourceTechnol*, (2010),101,7958-65.
- [8] Narayanan KB, Sakthivel "Phytosynthesis of gold nanoparticles using leaf extract of Coleus amboinicus Lour", *Mater Charact*, (2010), 61: 1232-1238.
- [9] Nair B, Pradeep T., "Preparation of gold nanoparticles from Mirabilis Jalapa Flower," *Crystal Growth Design*, (2002),2,293-298.
- [10] Schultz S, Smith DR, Mock JJ, Schultz DA. "Single-target molecule detection with nonbleachingmulticolor optical immunolabels. "*ProcNatlAcadSci*, (**2000**),97,996-1001.
- [11] O. Metz, W. Stoll, W. "Toxicity of cellular uptake gold nanoparticles" *Plenert, Cancer* (**1982**), 49(2),224-228.
- [12] Ratul Kumar Das1, Nayanmoni Gogoi1, PunuriJayasekhar Babu1, Pragya Sharma, Chandan Mahanta1, Utpal Bora1, "The synthesis of gold nanoparticles Amaranthusspinosus leaf extract and study of their optical proparties" Advances in Materials Physics and Chemistry, (2012), 2, 275-281.
- [13] Shankar SS, Rai A, Ankamwar B, Ahmad A, Sastry M. "Rapid synthesis of Au,Ag and bimetallic Au core Ag shell nanoparticles using (Azadirachtaindica)leaf broth." *J Colloid Interface Sci*, (2004), Vol. 275, No. 2, pp. 496-502.

- [14] Gardea-Torresdey JL, Parsons JG, Gomez E, Peralta-Videa J, Troiaz HE, Santiago P, Jose-Yacaman M. "Foemation and growth of Au nanoparticles inside live Alfalfa Plants" *Nano Letters*, (2002), 2: 397-401.
- [15] Ankamwar B, Damle C, Ahmad A, Sastry M, "Biosynthesis of gold and silver nanoparticles using EmblicsOfficialis Fruit extract and their Phase Transfer and Transmetaiiation in an Organic Solution" *J NanosciNanotechnol*, (2005).5,1665-71.
- [16] Huang J, Li Q, Sun D, Lu Y, Su Y, Yang X, Wang H, Wang Y, Shao W, He N, Hong J, Chen C. "Biosynthesis of silver and gold nanoparticles by novel sundried Cinnanonumcamphora leaf". *Nanotechnology*, (2007),18, 105-104.
- [17] Daizy Philip "Green synthesis of gold and silver nanoparticles using Hibiscus rosasinensis" *PhysicaE*.42, .(**2010**),pp.1417-1424.
- [18] Vilchis-Nestor A et al. "Solventless synthesis and optical properties of Au and Ag nanoparticles using Camellia sinensis extract ." *J Mater Lett.* (2008), Vol. 62,pp. 3103.
- [19] Narayana, K B., Sakthive, N *J* "Coriander leaf mediated biosynthesisnof gold nanoparticles" *Mater.Lett*, (**2008**) .. Vol. 68,pp.4588-4590.
- [20] Laura Castro "Biosynthesis of gold nanowires using sugar beat pulb". Process biochemistry. et al.(2011), 46,pp.1076-1082.
- [21] Mubarakali D "Plant extract mediated synthesis of silver and gold nanoparticles and its antibacterial activities against clinically isolated pathogens." *et al Colloids Surf. B.*(2011) ,.85,pp.360-365.
- [22] Alagumuthu, G., Kirubha, R.Inter. J. "Biofabrication and antibacterial activities of gold nanoparticles using AegleMarmelos fruit etract". *Adv. Sci and Technol.*(2013), .6(6),pp.73-95.



Available at https://edupediapublications.org/journals

p-ISSN: 2348-6848 e-ISSN: 2348-795X Volume 03 Issue 09 May 2016

- [23] Chopra RN, Nayar SL, Chopra IC, "Glossary of Indian Medicinal Plants". *Publication and Information Directorate*, CSIR, NewDelhi, 1992, 35, 241-245.
- [24] Nandkarni KM, "Indian MateriaMedica", *Popular Prakashan Pvt Ltd, Bombay,* 1995, 1,182.
- [25] Asima Chatterjee, Satyesh Chandra Pakrashi, "The Treatise on Indian Medicinal Plants", *Publications and Information Directorate*," *CSIR*, *New Delhi*, (1992), *Vol.*2 p.16 & 21.
- [26] Manoranjan Sharma H, Radhapyari Devi A, Manihan Sharma B, "Vegetable drugsused by the Meitei community of Manipur," *Indian Journal of Traditional Knowledge*, (2005), 4, 39-46.
- [27] Mukherjee PK, Gopal TK, Subburaju T, Dhanbal SP, Duraiswami B, Elango and Suresh B, "Studies on the antidiarrhoeal profiles of Bauhinia purpurea Linn Leaves fam. caesalpinaceae extract," *Natural Product Science*, (1998), 4, 234-237.
- [28] Pettit GR, Numata A, Iwamoto C, Usami Y, Yamada T, Ohishi H, Cragg GM, Antineoplastic agents. "Isolation and structures of bauhiniastatins 1-4 from Bauhinia purpurea," . *J Nat Prod*, (2006), 6, 323-7.
- [29] Panda S, Kar A, "Withaniasomnifera and Bauhinia purpurea in the regulation of circulating thyroid hormone concentrations in female mice", *J Ethno pharmacol*, (1999), 7,233-9.
- [30] Bauhinia purpurealinn.: Kumar T., Chandrashekar K.S. *Research Journal of Medicinal Plant* (2011), 5,pp.420-431.
- [31] T. R. Shantha, M. Prathapa Reddy, V. Rama Rao, Vandanabharthi, R. Kishore Kumar "Journal of Drug Research in Ayurvedic Sciences". (**2015**),(1)47:2272-0357.
- [32] Sharma OP, Aparna P, Neelu S, Pareek LK. " *Medico Ethno Botanically Important Plants of Jhunjhunu*", Rajasthan, Chapter-21, Ethno Medicinal

Plants of India (P.C.Trivedi), Aavishkars Publishers, Rajasthan, India, Ethno Medicinal Plants of India (**2007**),375, pp14.

- [33] Ramachandra, R. and Joshi BC. "Chemical examination of Bauhinia purpurea flower" *Current Science*. (1967),36,pp.pp 574-575.
- [34] Tiwari KP, Masood M, Rathore S, Minocha P. "Study of Anthocyanins from the flowers of some medicinal plants," , *VijnanaParishadAnusandhanPatrika*, (**1978**), vol.21 (2):177-178.
- [35] Bhartiya HP, Dubey P, Katiyar SB, Gupta PC. "A new chalcone glycoside from Bauhinia purpurea." *Phytochemistry* (1979), 18(4):pp 689.
- [36] M Iosin, P Baldeck, S Astilean, "Synthesis of gold nanoparticles by combining fluorescence and SERS spectroscopy". *J. Nanopart.* (2010) ,12,pp2843-2849.